MATERIAL SAFETY DATA SHEET

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Kocide® 101 Date Prepared: April 28, 1999

1. CHEMICAL PRODUCT AND COMPANY IDENTIFICATION

PRODUCT IDENTIFICATION

Product Name: Kocide® 101

HAZARD CLASSIFICATION (0-minimal, 1-slight, 2-moderate, 3-serious, 4-severe)

HMIS: HEALTH-1 FIRE-1 REACTIVITY-0

MANUFACTURER NAME AND ADDRESS

Griffin L.L.C. Griffin (Europe) S.A. Griffin FE (Malaysia) S/B

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EMERGENCY TELEPHONE NUMBERS

Griffin L.L.C. (USA): (+1) (800) 237 1854 Griffin (Europe) S.A.: (+32)-2-720 6644 Griffin FE (Malaysia) S/B: (+60)-3-757 4773 Prosar: (+1) (888) 324-7598 Chemtrec: (+1) (800) 424 9300

2. COMPOSITION/ INFORMATION ON INGREDIENTS

Component Name	% by Wt.	CAS#	ACGIH TLV	OSHA PEL
Copper Hydroxide	77.0	20427-59-2	1 mg/m ³ *	1 mg/m ³ *

^{*} As copper dusts or mists (CAS 7440-50-8).

Components not precisely identified are proprietary or not hazardous.

3. HAZARDS IDENTIFICATION

EMERGENCY OVERVIEW

Direct contact may seriously damage eye tissue. Slightly to non-toxic orally, dermally, and by inhalation. See below for route-specific details.

POTENTIAL HEALTH EFFECTS

Inhalation: Slightly toxic by inhalation. Excessive exposure may

cause cough, mucous production, shortness of breath,

reflecting metal fume fever.

Eye Irritation: Severely irritating to the eyes. Direct contact may

cause destruction of eye tissue. May be corrosive to

the eyes if not washed immediately.

Skin Irritation: Slight skin irritant. Excessive exposure, especially if

prolonged, may produce skin irritation. Repeated exposure may cause allergic contact dermatitis.

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Hazards Identification continued:

POTENTIAL HEALTH EFFECTS

Skin Absorption: Not a skin absorption hazard.

Ingestion: Slightly toxic by oral exposure. This material may

produce toxicity if ingested in large quantities.

Symptoms of over-exposure may include nausea and vomiting, abdominal pain, and central nervous system

depression.

Chronic: Low chronic toxicity unless excessive exposure is

encountered. Excessive exposure to copper by inhalation may result in irritation of the upper respiratory tract which, if severe, may lead to

perforation of the nasal septum after long periods of

exposure.

4. FIRST AID MEASURES

Inhalation: Remove victim to fresh air. If not breathing, give

artificial respiration preferably mouth-to-mouth. Get

professional medical attention immediately.

Eye Contact: Hold eyelids open and flush with water for 15-20

minutes until no evidence of chemical remains. Get

professional medical attention.

Skin Contact: Remove contaminated clothing and shoes. Wash with

plenty of soap and water for 15-20 minutes until no evidence of chemical remains. Get professional

medical attention.

Ingestion: Drink promptly a large quantity of milk, egg white,

gelatin solution or if these are not available, large quantities of water. Unless extensive vomiting has occurred, empty the stomach by gastric lavage with water, milk, sodium bicarbonate solution of a 0.1% solution of potassium ferrocyanide. (Gosselin, Clinical

Toxicology of Commercial Products, 5th Ed.).Administration of gastric lavage should be performed by qualified medical personnel. Probable mucosal damage may contraindicate use of gastric lavage.

Emergency Medical

Treatment: Treat symptomatically. Acute oral overexposure to

copper hydroxide, a major component of this product, may cause hypotension, hemolysis, and, rarely,

methemoglobinemia. Severe intoxication is associated

with serum copper levels greater than 500 mcg/dl.

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First Aid Measures continued:

Emergency Medical

Treatment: Copper hydroxide is an emetic, however, dilution with

fluids, adsorption with activated charcoal, or lavage may be indicated. Chelation therapy with BAL or D-penicillamine has proved useful in cases of acute

overexposure.

5. FIRE FIGHTING MEASURES

Flash Point & Method: Not determined Flammable Limits: Not determined Autoignition Temperature: Not determined

FIRE FIGHTING HAZARDS & PROCEDURES

General Hazard: Negligible fire hazard when exposed to heat or flame. Use dry chemical, carbon dioxide, water spray, or foam. Extinguishing Media:

No incompatible fire fighting media known.

Fire Fighting Instructions: Avoid contact with molten product to prevent serious

burns.

Wear protective clothing and self-contained breathing Fire Fighting Equipment:

apparatus.

Hazardous Combustion

Products: Decomposes to CuO and H₂0 above 140 F.

6. ACCIDENTAL RELEASE MEASURES

Land Spill: Sweep up and place in suitable (fiberboard) containers

for later disposal.

Water Spill: If feasible, copper may be precipitated/ultrafiltrated with

caustics or other chemicals and resulting sludge

disposed of in a chemical landfill.

7. HANDLING AND STORAGE

Store below 35° C (95° F). Average shelf life under Storage Temperature:

proper storage conditions is 2 years.

Ambient pressure. Storage Pressure:

General Information: Store in a clean, dry area. Do not store near feed, food

or within the reach of children.

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8. EXPOSURE CONTROLS/ PERSONAL PROTECTION

PESTICIDE APPLICATORS & WORKERS

These workers must refer to the Product Label and Directions For Use attached to the product for Agricultural Use Requirements in accordance with the EPA Worker Protection Standard 40 CFR part 170.

MANUFACTURING, COMMERCIAL BLENDING, & PACKAGING WORKERS

Ventilation: Control enclosed spaces with adequate ventilation to

prevent exceedance of ACGIH TLV or OSHA PEL (1

 mg/m^3).

Respiratory Protection: In enclosed spaces where the TLV or PEL may be

exceeded, wear NIOSH/MSHA approved dust or mist

respirator.

Eye Protection: Wear protective eyewear to prevent contact with this

substance.

Protective Clothing: Applicators and other handlers must wear long-sleeved

shirt, long pants, waterproof gloves, and shoes plus socks.

9. PHYSICAL AND CHEMICAL PROPERTIES

Vapor Pressure: Negligible vapor pressure

Vapor Density (Air=1): Not determined

Bulk Density: 102.86 - 154.29 g/m³

Specific Gravity: 8 - 12 lb/ft³

Solubility in Water: Disperses to form an emulsion. Copper hydroxide is

insoluble in cold water and decomposes in hot water.

pH: 8.9

Boiling Point:

Freezing Point:

Viscosity:

Not determined

Decomposes

Not determined

Odor: Characteristic copper odor

Color: Light blue Physical State: Powder

10. STABILITY AND REACTIVITY

General: This material is stable under normal conditions.

Incompatible Materials: Not determined Conditions to Avoid: Excessive heat.

Hazardous Decomposition: Decomposes to CuO and H₂0 above 140° F.

Hazardous Polymerization: Material is not known to polymerize.

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11. TOXICOLOGICAL INFORMATION

ACUTE

Inhalation: Acute inhalation $LC_{50} = 11.65 \text{ mg/L}$ (rat 1-hour). May cause

irritation of the mucous membranes. Exposure to copper fume may results in metallic taste, nausea, vomiting, and metal fume fever with chills, fever, aching muscles, dry

throat and headache.

Eye Irritation: Severely irritating to the eyes. May cause severe eye

irritation including permanent corneal opacity. May be

corrosive to the eyes if not washed immediately.

Skin Irritation: May cause slight irritation. Many copper salts cause itching,

eczema and, rarely, sensitization reactions in previously

exposed persons.

Skin Absorption: Acute dermal $LD_{50} > 5000$ mg/kg (rats). This product is

slightly toxic by dermal exposure.

Ingestion: Oral $LD_{50} = 833$ mg/kg (rats), indicating slight toxicity.

Ingestion of large doses of copper salts may result progressively in irritation of the gastrointestinal tract, nausea, vomiting, salivation, gastric pain, hemorrhagic gastritis, diarrhea, capillary damage, liver and kidney

damage, and central nervous system stimulation followed by depression. Jaundice, pain in the liver, and hemolytic anemia have been reported following acute human

poisonings.

CHRONIC: Repeated ingestion of copper salts may results in anemia,

liver, and kidney damage. Chronic inhalation exposure may cause a metallic taste in the mouth, irritation of the upper respiratory tract such as the nasal mucosa that may

progress to perforation of the nasal septum. Chronic cough may also occur. Copper hydroxide which comprises 77% of

this product governs the toxicity of the product. The remaining components have low to negligible toxicity.

Special Health Effects: Copper-intolerant individuals should not be exposed to this

material. No additional information is available on whether overexposure to this material would aggravate other existing

special medical conditions.

Toxicity of Individual

Components: This product contains 7 to 8% of an inert vehicle that, itself,

contains a small fraction of crystalline silica. Although crystalline silica is associated with silicosis and lung cancer, the International Agency for Research on Cancer (IARC) has evaluated this inert vehicle and designated it as Class 3

(insufficient evidence in either animals or humans to

conclude carcinogenic activity).

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12. ECOLOGICAL INFORMATION

Chemical Fate: The degree of mobility of copper in the environment

depends upon the pH of ambient soils and waters. The higher the acidity, the more soluble copper salts are and, hence, the more mobile. Partitioning of copper into air is negligible due to the low vapor pressure of copper

salts.

ECOTOXICITY (copper hydroxide)

Test Type	Species	Value
Aquatic LC ₅₀	Bluegill	180,000 ppb
Aquatic LC ₅₀ (96 hr)	Fathead Minnow	23 ppb
Aquatic LC ₅₀	Rainbow Trout	23 ppb
Aquatic EC ₅₀	Daphnia magna	6.5 ppb
Avian - acute oral LD ₅₀	Bobwhite Quail	>340 mg/kg
Avian - 8-day dietary LD ₅₀	Bobwhite Quail	>10,000 ppm
Avian - 8-day dietary LD ₅₀	Mallard Duck	>10,000 ppm

13. DISPOSAL CONSIDERATIONS

Comply with appropriate disposal regulations. Landfill solids at permitted sites. Use registered transporters.

14. TRANSPORT INFORMATION

Department of Transportation (DOT):

International Air Transport Association (IATA):

Not Regulated Not Regulated International Maritime Organization (IMO):

Not Regulated

15. REGULATORY INFORMATION

OSHA: This product is considered hazardous under the OSHA

Hazardous Communication Standard 29 CFR

§1910.1200.

TSCA: All product components are on the TSCA Chemical

Inventory.

CERCLA: Releases of a component of this material (metallic

copper, CAS No. 7440-50-8, RQ = 5000 lbs) to air, land, or water are reportable to the National Response Center under the Comprehensive Environmental Response, Compensation, and Liability Act (CERCLA) or to the state and local emergency planning committees under the Superfund Amendments and Reauthorization Act

(SARA) Title III Section 304.

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Regulatory Information contiuned:

RCRA: When a decision is made to discard this material as

supplied, it does not meet RCRA's characteristic definition of ignitability, corrositivity, or reactivity, and is

not listed in 40 CFR §261.33.

SARA TITLE III

311/312 Hazard Categories: This product has been reviewed according to the EPA

"Hazard Categories" and is categorized as an acute

health hazard (40 CFR §370.41).

313 Reportable Ingredients: This product contains a percentage of metallic copper

(CAS No. 7440-50-8) which is listed in Section 313

above de minimis concentrations.

STATE REGULATIONS

California: Listed under copper (CAS No. 7440-50-8) with footnotes

referencing above federal standards.

New Jersey: Footnotes reference above federal standards plus listing

of copper on the New Jersey Environmental Hazardous Substances List (NJ Department of Environmental Protection, Title 7 New Jersey Administrative Code

(NJAC) Chapter 1G).

Massachusetts: Listed as copper (CAS No. 7440-50-8) and

copper-based pesticide, solid, toxic (no CAS No.

assigned).

Pennsylvania: Listed as copper fume or dust (CAS No. 7440-50-8) as

an environmental hazard.

16. OTHER INFORMATION

REVISION SUMMARY

This Material Safety Data Sheet replaces the one dated 02/19/96 and has been revised using the standard Griffin ANSI Z400.1 compliant format. Revisions have been made in section 1 and in word processing format.

Kocide® is a registered trademark of Griffin Corporation.

The information in this Material Safety Data Sheet relates to this specific material. It may not be valid for this material if used in combination with any other materials or in any process. It is the users' responsibility to satisfy themselves as to the suitability and completeness of this information for their own particular use.